

Propeller-Shaped Fused Oligothiophenes: A Remarkable Effect of the Topology of Sulfur Atoms on Columnar Stacking

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Supporting Information

ABSTRACT: Propeller-shaped regioisomers of fused oligothiophenes **F9T_{endo}**, **F9T_{anti}**, and **F9T_{exo}** were successfully synthesized. DFT calculations indicated that their core parts are distorted from planarity due to intramolecular steric repulsions involving large sulfur atoms. In contrast with soft crystalline **F9T_{anti}** and **F9T_{exo}**, **F9T_{endo}** self-assembles into a hexagonal columnar liquid crystal (*Col_h* LC), displaying a clear X-ray diffraction (XRD) due to its stacked π -conjugated core. In each LC column, well-organized intermolecular S–S contacts are developed triple-helically along the columnar axis with a helical pitch of 4.04 nm. Among LC semiconductors reported to date, *Col_h* LC **F9T_{endo}** displays a top-class charge-carrier mobility ($0.18 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$) with a distinct ambipolar character featuring well-balanced hole and electron mobilities. A thin film, prepared by mixing **F9T_{endo}** with soluble fullerene PCBM, shows a photo-voltaic response, when the fullerene content is large enough to compensate a small absorptivity of **F9T_{endo}** for visible light.

With an increasing demand for organic electronics, particular attention has been focused on the development of new π -conjugated organic motifs and their elaborate assembly in the solid state.¹ Promising synthetic strategies along this direction include oxidative fusion of multiple aromatic rings to provide extended π -conjugated disks.² Hexabenzocoronene derivatives³ are representative of such disk-shaped aromatic motifs, where 13 benzene rings are fused two-dimensionally into a large π -conjugated disk. Meanwhile, in view of practical applications to electronic and optoelectronic devices, thiophene derivatives are quite attractive.^{4–6} However, reported examples of fused oligothiophene derivatives mostly adopt 2D tape and ribbon-like linear architectures,⁷ but their disk-shaped analogues are very limited.⁸ Although large thiophene-containing disks have been reported, they are prepared by fusion of triphenylene^{9a} and coronene^{9b} with thiophenes.

Here we report **F9T_{endo}**, **F9T_{anti}**, and **F9T_{exo}** (Figure 1) whose regioisomeric aromatic cores comprise nine nonlinearly connected fused oligothiophene units. We initially thought that these aromatic cores would adopt a disk shape appropriate for π -

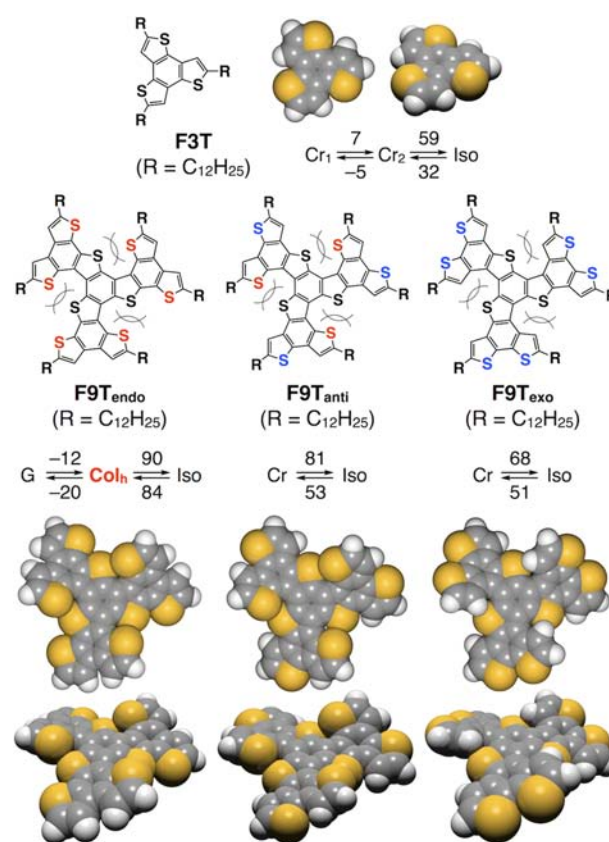


Figure 1. Chemical structures of fused oligothiophenes **F3T**, **F9T_{endo}**, **F9T_{anti}**, and **F9T_{exo}** (R = C₁₂H₂₅) together with their computer-generated aromatic core structures (R = H) and phase behaviors with transition temperatures (°C). The core structures were obtained by optimizing their methyl-substituted analogues (R = Me) using the density functional theory (DFT) at the B3LYP/6-31G(d) level. Symbols Cr, Col_h, Iso, and G in the phase behaviors denote crystalline, hexagonal columnar (*Col_h*) liquid crystalline, isotropic, and glassy phases, respectively.

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stacking and envisioned that F9Ts might self-assemble into columnar liquid crystalline (LC) materials with the help of their six dodecyl side chains. However, to our surprise, only F9T_{endo} forms a columnar LC assembly. We then noticed that the aromatic cores of F9Ts are not planar but adopt a C₃-symmetric propeller shape due to a considerable steric repulsion among proximal sulfur atoms. In the columnar assembly of F9T_{endo}, this particular structural feature allows for developing well-organized intermolecular S–S contacts¹⁰ if the stacked cores adopt a helical geometry.^{11,12} In this communication, we highlight such unique structural aspects and electronic properties of columnar assembled F9T_{endo}.

F9T_{endo}, F9T_{anti}, and F9T_{exo} were synthesized via FeCl₃-mediated oxidative cyclization of their bithiophene-substituted benzotrithiophene (BTT) precursors.¹³ Together with BTT-based F3T as a reference (Figure 1),⁸ the fused oligothiophenes thus obtained were unambiguously characterized by NMR spectroscopy, MALDI-TOF mass spectrometry, and elemental analysis.¹³ Because of their six dodecyl side chains, they are highly soluble in common organic solvents such as hexane, CHCl₃, and THF. As described in the introductory part, we initially thought that the fused oligothiophene cores of F9Ts, just as the case of F3T,⁸ adopt a planar conformation. However, structural optimization by DFT calculation (Figure 1) allowed us to notice that they are nonplanar, adopting a propeller shape with a C₃ symmetry. For example, in the case of F9T_{endo} and F9T_{anti}, three congested peripheral areas suffer from a considerably large steric repulsion between two proximal sulfur atoms (Figure 1). Although F9T_{exo} does not carry such proximal sulfur atoms, the same issue exists because even small Ar–H hydrogen atoms can sterically conflict with the proximal sulfur atoms. These steric issues do not allow the aromatic cores of F9Ts to be planarized.

As shown in Figure 4a (left), the electronic absorption spectra in THF of F9Ts, by reference to F3T (<306 nm), extend toward a long-wavelength region. Differential pulse voltammetry (DPV) of F9Ts in CH₂Cl₂, using NBu₄⁺PF₆[−] as a supporting electrolyte, gave shallow oxidation potentials as shown in Figure S8 (Supporting Information).¹³ Together with the HOMO–LUMO gaps estimated from the red absorption spectral edges, we evaluated the HOMO/LUMO energy levels as −5.6/−1.5 eV for F3T, −5.1/−2.0 eV for F9T_{endo}, −5.2/−2.0 eV for F9T_{anti}, and −5.3/−2.1 eV for F9T_{exo}. From these optical and electrochemical band gaps, one may notice that the persistent π -conjugation length of F9T_{endo} is slightly but definitely larger than those of the other two F9Ts. This is understandable, considering a previous report that α – α connections, involved in fused oligothiophene skeletons, are better for π -electronic conjugation than α – β and β – β connections.^{7e,14}

In differential scanning calorimetry (DSC), F9T_{anti} and F9T_{exo} upon heating only showed a transition from a crystalline phase to an isotropic melt (Figures 1 and S11, Supporting Information).¹³ However, F9T_{endo}, in contrast, displayed a LC mesophase over a wide temperature range from −12 to 90 °C (Figure S11, Supporting Information; from 84 to −20 °C upon cooling).¹³ Its crossed polarized optical microscope (POM) image showed a birefringent dendritic texture (Figure 2a) typical of columnar LC materials. In fact, upon exposure to a synchrotron radiation X-ray beam,¹⁵ F9T_{endo} in the LC state at 30 °C clearly displayed a set of diffraction peaks at, e.g., $q = 2.19, 3.81, \text{ and } 4.39 \text{ nm}^{-1}$ indexed as 100, 110, and 200 planes (Table S1, Supporting Information),¹³ respectively, of a hexagonal columnar (Col_h) lattice with a lattice parameter a (intercolumnar distance) of 3.31 nm. Noteworthy, the XRD pattern also displayed a distinct diffraction at $q = 17.3$

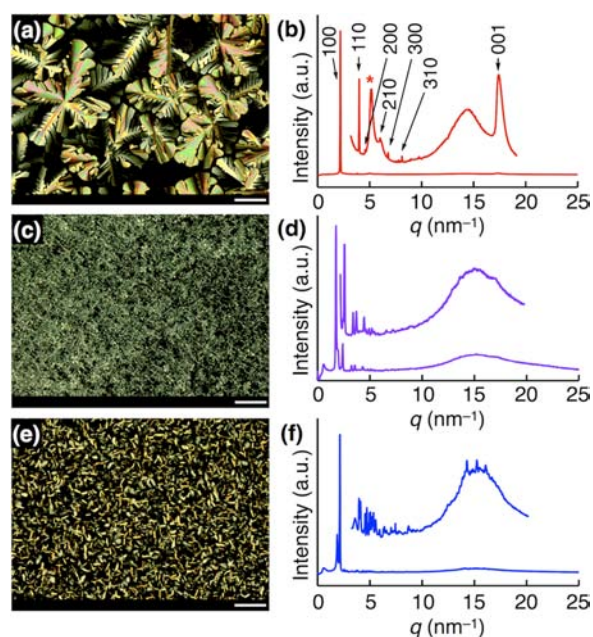


Figure 2. Crossed polarized optical micrographs (POMs) of (a) F9T_{endo} at 80 °C, (c) F9T_{anti} at 45 °C, and (e) F9T_{exo} at 64 °C on cooling from their isotropic melts at a rate of 0.5 °C min^{−1} and subsequent annealing for 2 h. Scale bars indicate 50 μm. X-ray diffraction (XRD) patterns of (b) F9T_{endo}, (d) F9T_{anti}, and (f) F9T_{exo} at 30 °C after cooling from their isotropic melts at a rate of 1 °C min^{−1} and subsequent annealing for 2 min. The q value for the asterisked diffraction in part b is 4.95 nm^{−1}. For detailed analytical conditions, see the Supporting Information.

nm^{−1} with a lattice parameter c of 0.364 nm, assignable to the stacking periodicity (Figure 2b). Why does the assembling behavior of F9T_{endo} differ from those of the other two regioisomers, despite the fact that no remarkable difference lies in their optical and electrochemical properties? Although this was a puzzling issue for us, we found that F9T_{endo} can take full advantage of S–S interactions¹⁰ for columnar stacking. A clue to solve this issue was given by a particular diffraction peak at $q = 4.95 \text{ nm}^{-1}$ (asterisked in Figure 2b) indicative of the presence of a long-range periodic structure. As described already, the π -stacked columns of F9T_{endo} adopt a helical chirality. For the purpose of obtaining a solid support for this geometrical feature, a sheared LC film of F9T_{endo} was prepared, whose brightness in POM showed an angle-dependent periodic change (Figure S12, Supporting Information).¹³ Furthermore, this sheared LC film, in grazing incidence in-plane X-ray diffraction (GIIXD) analysis (Figure S15, Supporting Information),¹³ gave different azimuthal plots for the diffractions indexed as $hkl = 001$ (Figure 3e) and 100 (Figure 3g), where the former showed two peaks at azimuthal angles (ϕ) = 0 and 180° relative to the sheared direction, whereas the latter exhibited two peaks at $\phi = 90$ and 270°. These features allowed us to confirm that the LC columns in the film are oriented along the sheared direction. Then, we examined the azimuthal plots (Figures 3f and S15, Supporting Information) for the asterisked diffraction in Figure 2b, which displayed two split peaks ($\pm 20^\circ$) centered at $\phi = 0$ and 180°. This result demonstrates that the columnar assembly of F9T_{endo} adopts a helical geometry along the columnar axis.¹² Taking into account the C₃ symmetry of F9T_{endo}, the helical pitch of the LC columns can be estimated as 4.04 nm.¹⁶ Since the stacking distance is 0.364 nm (*vide ante*), 11.1 molecules of F9T_{endo} are involved in a single pitch. Hence, the dihedral angle of two adjacent core units

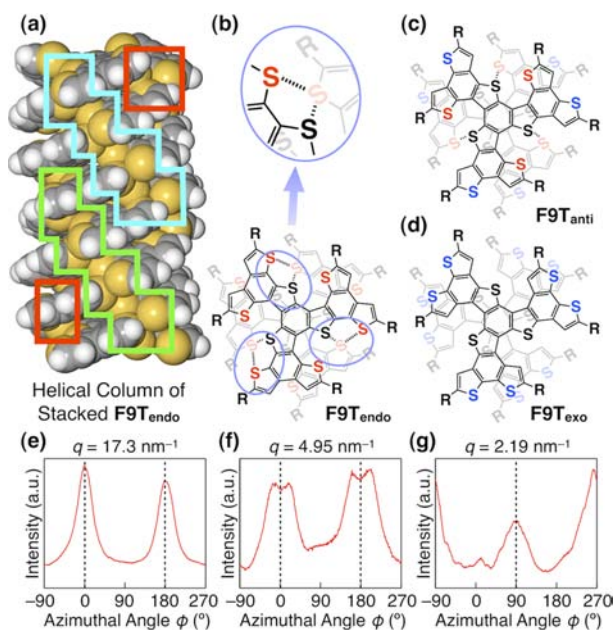


Figure 3. Schematic illustrations of (a) a columnar assembly of $F9T_{endo}$ via multiple S–S contacts adopting a triple-helical geometry and (b–d) stacked dimer models of (b) $F9T_{endo}$, (c) $F9T_{anti}$, and (d) $F9T_{exo}$ with a dihedral angle of 32.5° for investigating the availability of multiple S–S contacts. (e–g) Azimuthal plots for grazing incidence in-plane X-ray diffraction (GIIXD) intensities of a sheared LC film of $F9T_{endo}$ in regard to the X-ray diffractions at $q =$ (e) 17.3 ($hkl = 001$), (f) 4.95 (asterisked), and (g) 2.19 ($hkl = 100$) nm^{-1} in Figure 2b as a function of azimuthal angle (ϕ).

in a single column is calculated as 32.5° . Thus, in a stacked dimer model of $F9T_{endo}$ (Figure 3b), if the upper core is positioned with a dihedral angle of 32.5° onto the lower core, three sets of intermolecular S–S contacts are simultaneously made possible. Similar but certainly much weaker S–S contacts are expected for $F9T_{anti}$ (Figure 3c), whereas no such interactions are available for $F9T_{exo}$ (Figure 3d). In accord with this consideration, soft-crystalline $F9T_{anti}$ actually showed a weak diffraction peak at $q = 16.7 \text{ nm}^{-1}$ ($d = 0.376 \text{ nm}$; Figure 2d) assignable to the stacking distance of the core units.

With these structural features, we investigated solid-state electronic properties of $F9T_{endo}$, together with those of $F9T_{anti}$ and $F9T_{exo}$. As shown in Figure 4a, their visible-light absorptivities in the solid state are slightly better than those in THF. By means of photoelectron yield spectroscopy (Figure S16, Supporting Information),¹³ we evaluated the HOMO energy levels of $F9T$ s in the solid state. Along with the wavelengths for the red absorption spectral edges, their HOMO/LUMO energy levels were evaluated as $-5.0/-2.0 \text{ eV}$ for $F9T_{endo}$, $-5.1/-2.0 \text{ eV}$ for $F9T_{anti}$, and $-5.1/-1.9 \text{ eV}$ for $F9T_{exo}$. The observed energy levels in the solid state are rather close to one another but obviously differ from those of $F3T$ ($-5.6/-1.6 \text{ eV}$). Next, we investigated their carrier transport properties by means of flash-photolysis time-resolved microwave conductivity (FP-TRMC).¹³ FP-TRMC is a method that allows for evaluating short-range (intrinsic) carrier transport properties of materials without electrodes.¹⁷ In this method, a maximum transient conductivity is given by $\varphi \sum \mu_{\text{max}}$ where φ and $\sum \mu_{\text{max}}$ represent charge carrier generation efficiency and the sum of maximum hole and electron mobilities, respectively. When exposed to a 355 nm laser pulse, drop-cast films of $F9T_{endo}$, $F9T_{anti}$, and $F9T_{exo}$ on a quartz plate displayed a transient

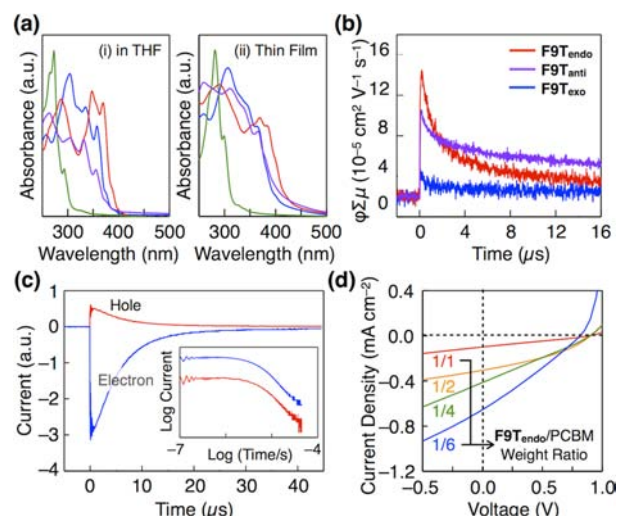


Figure 4. (a) Electronic absorption spectra of THF solutions (left; $1.0 \times 10^{-6} \text{ M}$) and spin-coated films (right) of $F3T$ (green), $F9T_{endo}$ (red), $F9T_{anti}$ (purple), and $F9T_{exo}$ (blue). (b) FP-TRMC profiles of cast films of $F9T_{endo}$ (red), $F9T_{anti}$ (purple), and $F9T_{exo}$ (blue) at 25°C upon excitation at 355 nm. (c) Time-of-flight photocurrent transients of $F9T_{endo}$ for hole (red) and electron (blue) at applied electric fields of $+3.3 \times 10^4$ and $-3.3 \times 10^4 \text{ V cm}^{-1}$, respectively. The inset shows log–log plots. (d) Current–voltage characteristics of BHJ solar cells using $F9T_{endo}$ /PCBM blends as active layers at their weight ratios of 1/1 (red), 1/2 (orange), 1/4 (green), and 1/6 (blue).

conductivity with prompt rise and slow decay features (Figure 4b).¹³ By taking into account the φ values obtained by the direct-current method (Figure S18, Supporting Information),^{17b} charge-carrier mobilities of $F9T_{endo}$, $F9T_{anti}$, and $F9T_{exo}$ were evaluated as 0.18, 0.13, and $0.05 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, respectively. The mobility of $0.18 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$ for $F9T_{endo}$ is one of the largest among those ever reported for LC semiconductors.^{11,17a,18} Although soft-crystalline $F9T_{anti}$ and $F9T_{exo}$ are both inferior to Col_h LC $F9T_{endo}$, the former with a π -stacking profile in XRD (Figures 2d) is better than the latter having no distinct stacking feature (Figure 2f). Being encouraged by the large charge-carrier mobility of $F9T_{endo}$, we evaluate its long-range charge-carrier mobility by a time-of-flight technique.¹³ Of particular interest, this LC material displayed a distinct ambipolar carrier transport character along with an excellent conducting profile (Figure 4c). For example, at an applied electric field of $3.3 \times 10^4 \text{ V cm}^{-1}$, the hole and electron mobilities were 0.02 and $0.03 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$, respectively, and well balanced (Figure S19, Supporting Information).¹³

These observations prompted us to investigate whether $F9T_{endo}$, in combination with a soluble fullerene (phenyl C_{61} butyric acid methyl ester; PCBM), is usable as an active layer for a bulk-heterojunction organic photovoltaic (OPV) cell. Most likely due to an intrinsically small absorptivity of $F9T_{endo}$ for visible light (*vide ante*), the power conversion efficiency (PCE) at an $F9T_{endo}$ /PCBM ratio of, e.g., 1/1 (w/w), was only 0.023% (Figure 4d). However, by increasing the content of visible-light absorbing PCBM, the efficiency was improved to some extent, where a PCE value of 0.15% was obtained at an $F9T_{endo}$ /PCBM ratio of 1/6 (w/w).¹⁹ This result encourages further studies to enhance the visible-light absorptivity of the LC material.

In conclusion, we developed a new family of fused oligothiophenes $F9T$ s (Figure 1), which adopt a C_3 -symmetric propeller geometry. $F9T_{endo}$ self-assembles into a hexagonal

columnar LC mesophase over a wide temperature range including room temperature. Its LC columns are constructed by well-organized intermolecular S–S contacts that develop triple-helically along the columnar axis with a pitch of 4.04 nm (Figure 3). So far, S–S interactions have mostly been utilized for enhancing the 2D structural integrity of crystalline materials but are much less explored for the formation of columnar assembled π -conjugated motifs.^{10b} Among LC semiconductors reported to date, F9T_{endo} shows a top-class charge-carrier mobility with a well-balanced ambipolar character (Figure 4). Structural elaboration of F9T_{endo} to enhance its visible-light absorptivity is one of the interesting subjects worthy of further investigation.

■ ASSOCIATED CONTENT

Supporting Information

Details of synthesis and characterization of F3T and F9Ts, NMR, POM, XRD, electrochemical, and electrical data, and results of computational simulations. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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(19) An OPV device composed of F9T_{endo}/PCBM (1/6 w/w) showed $J_{SC} = 0.65 \text{ mA cm}^{-2}$, $V_{OC} = 0.81 \text{ V}$, FF = 0.27, and η (PCE) = 0.15% (Figure 4d).